

Sulphuric Acid 98%

Chemical Formula: H₂SO₄

Properties:

Appearance: Colourless liquid

Sp.Gravity at 30 °C: 1.82 Boiling point: 340 °C

Attack many metals with liberation of hydrogen, which is inflammable and forms explosive mixture, Hygroscopic in nature, Oxidizing agent.

Specifications:

 Purity (as H₂SO₄) :
 98.0% min

 Iron (as Fe) :
 0.05% max

 Residue on Ignition :
 0.20% max

 Lead (as Pb) :
 0.005% max

 Arsenic (as As) :
 0.004% max

Applications:

Sulphuric acid is an important reagent in a large number of processes. Among the many uses, some of the main ones are :

- Catalyst and dehydrating agent in many organic chemical manufacturing and petrochemical processes.
- In the manufacturing of detergents.
- In the manufacturing of titanium dioxide (a widely used white pigment), and other dyes and pigments.
- In the manufacturing of fertilizers mainly phosphates but also zinc, ammonium and potassium sulphates.
- In the manufacturing process for a range of plastics.
- Metal processing such as picking and de-scaling of steel, and non-ferrous metal plating and purification.

Some of its other uses include:

• Drug production via sulphonation, manufacturing of food acids (citric and lactic acid) and edible oils, Adhesives, Explosives, Synthetic rubber, Water and effluent treatment, Chlorine drying, Wood pulping, Leather tanning, Batteries.

DANGERS AND PRECAUTIONS FOR HANDLING PRODUCT:

- Toxic
- Causes severe damage to eyes, skin & air passage
- Reaction with moist air produces mist, which has severe irritant effect on eyes, skin & air passage.
- Store the material in MS acid proof brick lined or HDPE tank.
- Fire risk in contact with organic substances.



Packaging:

Tanker Load. HM HDPE drums of 235 kg & HM HDPE Carboys of 50 kg.

Product List:

Dimethyl Sulphate | Sulphuric Acid 70 % | Aluminum Sulphate (Non Ferric Alum) | Oleum 23% / 65% |
Battery Grade Sulphuric Acid | Sulphur Trioxide (SO3) | Chloro Sulphonic Acid | Sulfamic Acid